

SCHISANLACTONE A, A NEW TYPE OF TRITERPENOID FROM A SCHISANDRA SP.

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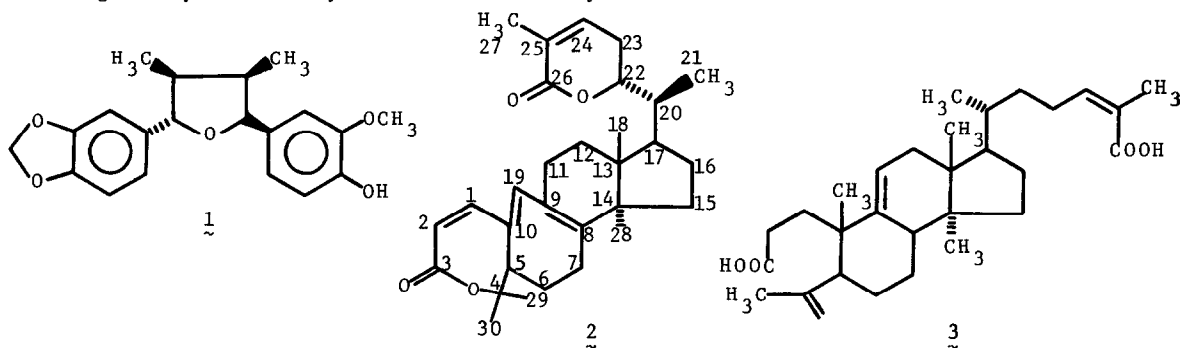
Abstract: Schisanlactone A, a triterpenoid isolated from a Schisandra sp., is shown to possess structure 2. The structure was assigned on the basis of spectroscopic studies and was confirmed by X-ray diffraction studies. Schisanlactone A represents a new structural type among the triterpenoids.

Recently, the structure of chicanine (1), the first tetrahydrofuran lignan isolated from the Schisandra genus, was reported.¹ Further separation of the crude extract of this unidentified Schisandra sp., collected in the Xiangfan district of the Hubei province of China, has provided a series of triterpene lactones named schisanlactone A, B, C, etc.² In this paper, we wish to report the structure of schisanlactone A, a new type of triterpenoid.

Schisanlactone A (2), C₃₀H₄₀O₄ (M⁺ 464.2954), mp 227—229°C, [α]_D²³ +365° (c 0.20, CHCl₃), displays two carbonyl bands (1725, 1685 cm⁻¹) but no hydroxyl absorption in the IR. The ¹³C NMR spectrum reveals carbonyl carbons at δ 166.93 and 166.28 and oxygenated carbons at δ 80.37 (s) and 80.24 (d). Thus schisanlactone A possesses two lactone rings, one tertiary and one secondary. Since kadsuric acid (3) has been isolated from a plant of the same genus,³ we considered that the lactone rings might arise from similarly located carboxyl groups. Indeed, ¹H-¹H selective decoupling experiments indicate the presence of a six membered α,β-unsaturated lactone ring. On irradiation of H-23A and H-23B, the signal of H-22 was decoupled to a doublet and the signal of H-24 to a quartet. On irradiation of H-22, the signal of H-24 was unaffected, whereas when the methyl protons (H-27) were irradiated, the signal for H-24 appeared as a doublet.

doublet. Thus, the side chain of schisanlactone A contains an α -methyl- α,β -unsaturated δ -lactone unsubstituted at the γ -position.

The ^1H NMR and ^{13}C NMR spectra show that schisanlactone A has only six methyl groups, one quaternary methyl less than kadsuric acid, and there is no evidence for the presence of a cyclopropane ring in the molecule. Furthermore, the UV spectrum, $\lambda_{\text{max}}^{\text{MeOH}}$ nm (log ϵ): 334(4.25), 265(3.89), 212(4.11), indicates that the three remaining double bonds are conjugated with the other lactone carbonyl.⁴ Thus schisanlactone A appeared to possess an A,B-seco-9,19-cyclo-lanostene skeleton, since only in this way, and with cleavage of the 9,10-bond, is it possible to construct the required extended system of conjugation to the A ring lactone. Structure 2, derived in this manner is compatible with the remaining ^1H NMR and ^{13}C NMR data.⁵ In order to verify the proposed structure, a single crystal X-ray diffraction study was carried out on schisanlactone A.



Clear, light yellow crystals of schisanlactone A were obtained from CH_2Cl_2 : MeOH (3:1). The crystal system was monoclinic, lattice constants $a = 12.101$ (1), $b = 14.195$ (1), $c = 7.8238$ (5) Å, and $\beta = 81.600$ (6)°, space group $P2_1$ with one molecule forming the asymmetric unit. A phasing model was achieved using a direct methods phasing approach.⁶ Hydrogens were located on a difference synthesis following partial refinement. Block-diagonal least squares refinement using anisotropic non-hydrogen atoms and isotropic hydrogens has converged to a standard crystallographic residual of 0.042 for the observed reflections.⁷ A perspective view of the final X-ray model of schisanlactone A (2) less hydrogens is shown in Figure 1. The X-ray experiment did not define the absolute configuration and this is based on the CD measurements discussed below. The stereochemical designations are 5(R), 13(R), 14(R), 17(S), 20(S), and 22(R). The cyclopentane ring is in the half-chair (C_2) conformation and the cyclohexene ring is in the twist boat conformation. The A-ring is aligned at an approximate dihedral angle of 45° to the CD ring system. This twist is mediated by the twist boat conformation of the B-ring. The C8-C9-C19-C10 dihedral angle is approximately 40° so that the conjugation of the third double bond is diminished.

Schisanlactone A shows a strongly positive Cotton effect at 244 nm ($\Delta\epsilon +7.4$), similar to ajugalactone⁸ and jaborosalactone A⁹ which have similar lactone

moieties in the side chain, and thus C-22 is assigned the (R)-configuration.¹⁰ To the best of our knowledge the carbon skeleton of schisanlactone A has not been previously observed among the triterpenoids.¹¹

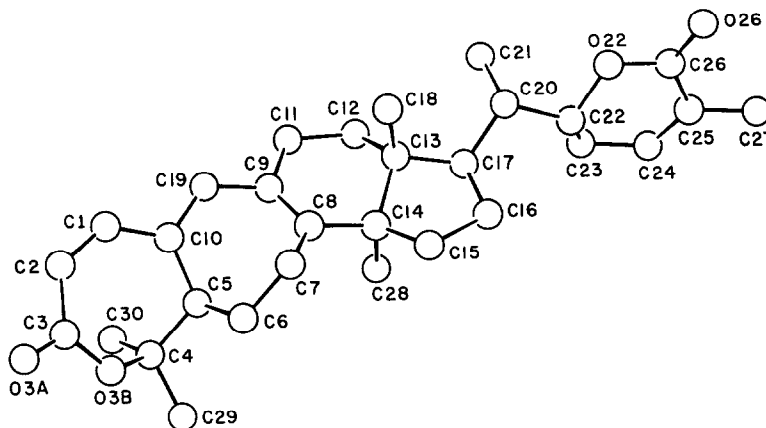


Figure 1. A computer generated perspective drawing the final X-ray model of schisanlactone A less hydrogens.

References and Notes

1. Jia-Sen Liu, Mei-Fen Huang, Yao-Liang Gao, and John A. Findlay, *Can. J. Chem.*, **59**, 1680 (1981).
2. Schisanlactone A elutes directly after chicanine during chromatography over silica gel (see reference 1) and is followed by schisanlactone B, C, etc.
3. Jia-Sen Liu, Mei-Fen Huang, and Yao-Liang Gao, *Acta Chim. Sin.*, **38**, 361 (1980).
4. A.I. Scott. Interpretation of the Ultraviolet Spectra of Natural Products. Pergamon Press, Oxford, 1964, p. 79.
5. Complete Table of ¹H NMR and ¹³C NMR data are available from W.A.A.
6. All crystallographic calculations were done on a PRIME 850 computer, operated by the Cornell Chemistry Computing Facility. Principal programs employed were REDUCE and UNIQUE, data reduction programs: Leonowicz, M.E., Cornell University, 1978; MULTAN78, 'A System of Computer Programs for the Automatic Solution of Crystal Structures from X-ray Diffraction Data,' direct methods programs and Fast Fourier Transform routine (locally modified to perform all Fourier calculations including Patterson syntheses): Main, P.; Hull, S.E.; Lessinger, L.; Germain, G.; Declercq, J.-P.; Woolfson, M.M., University of York, England, 1978; NQUEST, CYBER 173 version, negative quartets figure of

- merit calculation: Weeks, C.M., Medical Foundation of Buffalo, Inc., August 1976; BLS78A, anisotropic block-diagonal least squares refinement: Hirotsu, K.; Arnold, E., Cornell University, 1980; ORTEP, crystallographic illustration program: Johnson, C.K., Oak Ridge, ORNL-3794; For a summary description of MULTAN, see: Germain, G.; Main, P.; Woolfson, M.M., Acta Crystallogr. Sect. B, 1970, B26, 274-285. Woolfson, M.M., Acta Crystallogr. Sect. A, 1977, A33, 219-225. For a summary of NQEST, refer to: De Titta, G.T.; Edmonds, J.W.; Langs, D.A.; Hauptman, H.; Acta Crystallogr. Sect. A, 1975, A31, 472-479.
7. Tables of crystallographic data for schisanlactone A are available from the Cambridge Crystallographic Data Centre, University Chemical Laboratory, Lensfield Road, Cambridge CB2 1EW and from J.C.
 8. M. Koreeda, K. Nakanishi, and M. Goto, J. Am. Chem. Soc., 92, 7512 (1970).
 9. R. Tschesche, H. Schwang, H.-W. Fehlhaber and G. Snatzke, Tetrahedron, 22, 1129 (1966).
 10. G. Snatzke, H. Schwang, and P. Welzel in "Some Newer Physical Methods in Structural Chemistry," R. Bonnett and T.G. Davis, Ed., United Trade Press Ltd., London, 1967, p. 159.
 11. This work was supported by the National Institutes of Health (J.C.) and the Natural Sciences and Engineering Research Council of Canada (W.A.A.).

(Received in USA 3 January 1983)